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IN THE CLAIMS:

Amend the claims as follows:

- 1. (Original) A spray pyrolysis method, characterized in that it is applied to the synthesis of nanoparticles with a closed structure of metal chalcogenides having a lamellar crystallographic structure, of the general formula M_aX_b , in which M represents a metal and X a chalcogen, a and b representing the respective proportions of metal and of chalcogen, and in that it comprises pyrolysis of a liquid aerosol obtained from a solution of at least one precursor of a metal (M) and of a chalcogen (X), or of at least one precursor of said metal (M) and of at least one precursor of said chalcogen (X), dissolved in a solvent, said solution being atomized into fine droplets in suspension in a carrier gas.
- 2. (Original) The method as claimed in claim 1, characterized in that it comprises the following steps:
- formation of a solution of said at least one precursor of a metal and of a chalcogen, or of said at least one precursor of said metal and of said at least one precursor of said chalcogen in a solvent,
- atomization of said solution in liquid aerosol form by a nebulizer, in particular of the pneumatic or ultrasonic type, through which the carrier gas is flowing,

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- injection of the aerosol into a heated furnace to evaporate the solvent and to react and/or break down said precursor(s) of the metal and of the chalcogen so as to form the nanoparticles,
 - transport by the carrier gas of the nanoparticles to the furnace outlet, and
 - recovery of the nanoparticles at the furnace outlet.
- 3. (Currently Amended) The method as claimed in claim 1 [[or 2]], characterized in that said precursor of the metal and of the chalcogen contains both the metal and the chalcogen.
- 4. (Original) The method as claimed in claim 3, characterized in that said precursor is of the formula $(A)_cM(X)_d$ in which A is a cation such as K^+ , Na^+ or NH_4^+ , M is a metal and X a chalcogen, c and d respectively representing the number of cations and chalcogens.
- 5. (Currently Amended) The method as claimed in any one of claims 1 to 4 claim 1, characterized in that said metal is a transition metal selected from among Ti, Zr, Hf, V, Nb, Ta, Mo, W, Re, Co, Ni, Pt, Pd, Cr and Ru.
- 6. (Currently Amended) The method as claimed in any one of claims 1 to 4 claim 1, characterized in that said metal belongs to group III of the Periodic Table of Elements, such as Ga and In.

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- 7. (Currently Amended) The method as claimed in any one of claims 1 to 4 claim 1, characterized in that said metal is a metal from group IV of the Periodic Table of Elements, in particular Sn, Pb or Ge.
- 8. (Currently Amended) The method as claimed in any one of claims 1 to 4 claim 1, characterized in that said metal is a metal from group V of the Periodic Table of Elements, such as Bi.
- 9. (Currently Amended) The method as claimed in any one of claims 1 to 8 claim 1, characterized in that the chalcogen is selected from among oxygen, sulfur, selenium or tellurium.
- 10. (Currently Amended) The method as claimed in either of claims 4 or 9 claim4. characterized in that said precursor is a tetrathiometallate or a tetraselenometallate.
- 11. (Original) The method as claimed in claim 10, characterized in that the metal is molybdenum or tungsten.
- 12. (Currently Amended) The method as claimed in any one of claims 1 to 11 claim 1, characterized in that said carrier gas is an inert gas selected from nitrogen and argon and/or hydrogen.
- 13. (Currently Amended) The method as claimed in any one of claims 1 to 12 claim 1, characterized in that said solvent is a polar solvent, in particular water and/or ethanol.

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14. (Currently Amended) The method as claimed in any one of the preceding claims claim 1, characterized in that said nanoparticles are nanotubes, fullerenes and/or nanoboxes.

15. (Original) Nanoparticles of metal chalcogenides MX₂, characterized in that they have the form of nanoboxes made up of closed, generally hollow right parallelepipeds and rectangles.